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A Novel Phenol-Forming Reaction for Preparation of Benzene, Furan, and Thiophene Analogs of CC-1065/Duocarmycin Pharmacophores

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Abstract: A palladium-catalyzed intramolecular coupling reaction between methyl ketones and triflates was developed for preparation of phenol derivatives 13, 14, and 15 from 10, 11, and 12 in high yields. Potent antitumor substances 35, 36, and 37 were prepared from 13, 14, and 15.

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In the previous communication, we reported a new synthetic pathway for duocarmycin SA (1), ^{2, 3} an extremely potent antitumor antibiotic isolated from a culture broth of *Streptomyces* species (Scheme 1). ⁴ This synthesis featured a ready access to the key compound 9 having a pyrroloquinolinol structure, whose pyridine part was utilized for construction of the cyclopropapyrrolidine unit in 1. Preparation of 9 was accomplished by applying a palladium-catalyzed reaction, originally described by Kuwajima and co-workers, ⁵ to a silyl enol ether 6 of the triflate 5, derived from 2 and 3 (Stille coupling to 4 and further functional modification). Reaction conditions shown below afforded a cyclized product as a mixture of valence tautomers 7 and 8, and this was isolated in the form of methyl carbonate 9 in 89% yield calculated from 5. However, a high yield result was only limited to this case, and when these conditions were applied to the cyclizations of benzene, furan, and thiophene derivatives 10, ⁶ 11, ⁶ and 12, ⁶ the expected tricyclic compounds 13, 14, and 15 were obtained in poor to

Scheme 1

MeOOC
$$\stackrel{N}{H}$$
 $\stackrel{O}{O}$ $\stackrel{N}{H}$ $\stackrel{OMe}{O}$ $\stackrel{Br}{H}$ $\stackrel{R}{O}$ $\stackrel{N}{M}$ $\stackrel{N}{O}$ $\stackrel{N}{M}$ $\stackrel{N}{M}$ $\stackrel{N}{O}$ $\stackrel{N}{M}$ $\stackrel{N}{$

moderate yields of 24%, 27%, and 70%, respectively. In this paper, we report alternative reaction conditions applicable to these cyclizations in general.

For the first trial, the above palladium-catalyzed reaction was carried out by heating of furyl methyl ketone 11 in xylene in the presence of NaH without addition of Bu₃SnF and LiCl (Table, run 1). The desired product 14 was obtained in 9% yield, along with a variety of by-products 16, 17, and the pyridone derivative derived from 11, after methoxycarbonylation. Change of the base from NaH to t-BuOK or K₂CO₃ completely retarded the formation of 14, and only the above by-products were obtained in poor yields (run 2, 3). Use of Cs₂CO₃ effected the production of 14 in 26% yield, together with much decreased formations of 16 and 17 in 6% and 2% yields (run 4). Addition of Ph₃P dramatically enhanced the production of 14, and finally, the reaction conditions shown in run 5, i.e., use of 10 mol% of PdCl₂(Ph₃P)₂, 3 eq. of Cs₂CO₃, 0.3 eq. of Ph₃P in refluxing benzene under an Ar atmosphere, afforded the desired 14 in 90% yield without any by-product. When these reaction conditions were applied to 10 and 12, high yield formation of 13 and 15 was also noticed as shown in runs 6 and 7. Therefore, a novel type of phenol-forming reaction has been developed using a palladium catalyst, and will be widely

TfO N TfO N base
$$CICO_2Me$$
 NaH NaH

Run	Starting material	Pd cat.	Base (eq.)	Additive (eq.)	solvent	time h	Product	By-product	Recovery
1	11	0.05	NaH (3)		xylene	16.5	14, 9%	16, 11% 17, 6%	19%
2	11	0.05	t-BuOK (3)		toluene	16	14, 1%	pyridone, 13% 16, 3%	5%
3	11	0.05	K ₂ CO ₃ (3)	_	xylene	18		16, 18% 17, 3% pyridone, 16%	25%
4	11	0.05	Cs ₂ CO ₃		xylene	4.5	14, 26%	16, 6% 17, 2%	6%
(5)	11	0.1	Cs ₂ CO ₃ (3)	Ph ₃ P (0.3)	benzene	6	14, 90%		
6	10	0.1	Cs ₂ CO ₃ (3)	Ph ₃ P (0.3)	benzene	1.5	13, 91%		_
7	12	0.1	Cs ₂ CO ₃ (3)	Ph ₃ P (0.3)	benzene	3	15, 86%	_	· —

applicable to similar instances.

A precise reaction mechanism is unclear at this moment (Scheme 2). Presumably the main path A for generation of 21 starts with an intramolecular nucleophilic attack of the enolate to the palladium complex part in 19, resulting in a cyclic palladium species 20. This would then afford 21 on reductive elimination of Pd(0), which returns back to the reaction cycle. In some cases, 11 behaves as a sulfonating agent and 16 is partially formed from 21 in the presence of a base. For the formation of 17, our speculation is shown in path B. Intramolecular Heck reaction in 19 will afford 22. Then positional exchange of the palladium might take place by way of 23 to give 24, and further elimination of 25 from 24 would afford 17. Addition of Ph₃P may increase population of 19 bearing ligands on the palladium, and not only accelerates the path A, but also prevents the reaction course from proceeding to the path B, due to the steric hindrance in the Heck reaction.

As the requisite quinolinol derivatives 13, 14, and 15 became to be obtained in high yields, preparation of the key compounds 32, 2c, 7 33, 2c, 8, 9 and 34, 9 for benzene, furan, and thiophene analogs having CC-1065/duocarmycin pharmacophores, was carried out in a straightforward manner according to the previous report (Scheme 3). Treatment of 13, 14, and 15 with NaBH₄ in the presence of ClCO₂Me gave mixtures of 1,2- and 1,4-dihydropyridine derivatives with the CO₂Me group at the nitrogen atom. Without separation, these were dihydroxylated by oxidation with a catalytic amount of OsO₄ in the presence of Me₃NO, followed by reduction with Et₃SiH and BF₃·OEt₂. The hydroxyl groups adjacent to the nitrogen and the aromatic ring were reductively eliminated, and 26, 27, and 28 were obtained in 68%, 67%, and 60% yields, respectively. These were mesylated,

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36: X = O 37: X = S

and 29, 30, and 31, obtained in 99%, 94%, and 95% yields, were treated with K_2CO_3 in MeOH at room temperature for 1.5-4 h to afford 32, 33, and 34 in 94%, 88%, and 93% yields. Potent antitumor compounds 35, 2c , 10 36, 9 and 37, were prepared in 88%, 74%, and 63% yields from 32, 33, and 34 by condensation with imidazolide 38 using K_2CO_3 as a reagent in DMF at room temperature. This type of reaction, N-acylation of the vinylogous amide, is usually carried out by treatment with NaH in DMF. In our cases, the K_2CO_3 method gave some times superior results to the conventional procedure. 11

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